

Subject card

Subject name and code	Inorganic chemistry, PG_00103620						
Field of study	Environmental Protection						
Date of commencement of studies	October 2026		Academic year of realisation of subject		2026/2027		
Education level	Bachelor's studies		Subject group		Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	full-time studies		Mode of delivery		at the university		
Year of study	1		Language of instruction		Polish		
Semester of study	2		ECTS credits		2.0		
Learning profile	academic		Assessment form		exam		
Conducting unit	Laboratory of Physicochemistry of Coordination Complexes -> Department of General and Inorganic Chemistry -> Faculty of Chemistry -> Rector						
Name and surname of lecturer (lecturers)	Subject supervisor		dr hab. Joanna Makowska				
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	0.0	0.0	0.0	15
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	15		2.0		33.0	50
Subject objectives	familiarization with the basic types of inorganic compounds and methods of balancing chemical reaction equations introduction to the basics of chemical calculations developing the ability to independently experiment and solve problems when conducting a chemical experiment.						
Learning outcomes	Course outcome		Subject outcome		Method of verification		
	[OŚL3_W01] Discusses the basic concepts of mathematics, physics, chemistry and biology. Describes physical, chemical and biological phenomena occurring in nature as well as geological, geomorphological and climatic conditions of the functioning of nature.		It discusses at an advanced level the concepts of mathematics, physics, chemistry and biology, describes the physical, chemical and biological phenomena occurring in nature as well as the geological, geomorphological and climatic conditions of the functioning of nature.		[SW4] test/exam - oral or written		

Subject contents	Course contents		
	Topics of the lecture: periodicity and the chemistry of the elements, physicochemical properties of inorganic and coordination compounds. The following items are included: periodicity, chemical bonding, coordination compounds, types of chemical reactions, properties of chemical elements and their compounds. The groups of elements are presented in the following order: group 1, group 2, group 13, group14, group 15, group 16, group 17, group 18, and d-elements (groups 3-12; first transition row, second transition row, and third transition row).		
	Topics of auditory classes: basic types of inorganic compounds, balancing redox reactions, equilibria in the solutions of electrolytes.		
	Topics of lab classes: investigation of physicochemical properties of the elements, inorganic and coordination compounds based on chemical experiments.		
Prerequisites and co-requisites			
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
		51.0%	100.0%
Recommended reading	Basic literature	A. Literatura wymagana do ostatecznego zaliczenia zajęć (zdania egzaminu): A.1. wykorzystywana podczas zajęć Praca zbiorowa Obliczenia z chemii ogólnej - skrypt UG Praca zbiorowa Ćwiczenia laboratoryjne z chemii ogólnej. I. Część teoretyczna - skrypt UG Praca zbiorowa Ćwiczenia laboratoryjne z chemii ogólnej. II. Część doświadczalna - skrypt UG	
	Supplementary literature	A.2. studiowana samodzielnie przez studenta A. Bielański Podstawy chemii nieorganicznej J. D. Lee Zwięzła chemia nieorganiczna L. Jones, P. Atkins Chemia ogólna B. Literatura uzupełniająca L. Pajdowski Chemia ogólna M. J. Sienko, R. A. Plane Chemia. Podstawy i zastosowania	
	eResources addresses		
Example issues/ example questions/ tasks being completed	brak		
Work placement	Not applicable		

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